

# COMPENDIUM OF INTERNATIONAL METHODS OF ANALYSIS FOR SPIRITUOUS BEVERAGES AND ALCOHOLS

## DETERMINATION BY ISOTOPE RATIO MASS SPECTROMETRY OF THE $^{13}\text{C}/^{12}\text{C}$ RATIO OF WINE

ETHANOL OF SPIRIT DRINKS OF VITIVINICULTURAL ORIGIN. (Type II)

### **OIV-MA-BS-22 Determination by isotope ratio mass spectrometry**

## **of the $^{13}\text{C}/^{12}\text{C}$ ratio of wine ethanol of spirit drinks of vitivinicultural origin**

Type II method

### **1. Field of application**

The method enables the measuring of the  $^{13}\text{C}/^{12}\text{C}$  isotope ratio of the ethanol of spirit drinks of vitivinicultural origin.

### **2. Reference standards**

ISO 5725 :1994 «Accuracy (trueness and precision) of measurement methods and results: Basic method for the determination of repeatability and reproducibility of a standard measurement method»

V-PDB : Vienna-Pee-Dee Belemnite ( $R_{\text{PDB}} = 0.0112372$ ).

Method OIV «Detection of enriching musts, concentrated musts, grape sugar and wine by application of nuclear magnetic deuterium resonance (SNIF-NMR)»

### **3. Terms and definitions**

$^{13}\text{C}/^{12}\text{C}$  : Carbon 13 and carbon 12 isotope ratio for a given sample

$\delta^{13}\text{C}$  : Carbon 13 contents ( $^{13}\text{C}$ ) expressed in parts per 1000 (‰)

SNIF-NMR: Site-specific natural isotope fractionation studied by nuclear magnetic resonance

V-PDB : Vienna-Pee-Dee Belemnite. or PDB, is the main reference for measuring natural variations of carbon 13 isotopic contents. Calcium carbonate comes from a Cretaceous belemnite from the Pee Dee formation in South Carolina (USA). Its isotopic ratio  $^{13}\text{C}/^{12}\text{C}$  or  $R_{\text{PDB}}$  is  $R_{\text{PDB}} = 0.0112372$ . PDB reserves have been exhausted for a long time, but it has remained the primary reference to express natural variations of Carbon 13 isotopic contents. Reference material is calibrated based on this content and is available at the International Agency of Atomic Energy (IAEA) in Vienna (Austria). The isotopic indications of naturally occurring carbon 13 are expressed by V-PDB, as is the custom.

m/z: Mass to charge ratio

# COMPENDIUM OF INTERNATIONAL METHODS OF ANALYSIS FOR SPIRITUOUS BEVERAGES AND ALCOHOLS

## DETERMINATION BY ISOTOPE RATIO MASS SPECTROMETRY OF THE $^{13}\text{C}/^{12}\text{C}$ RATIO OF WINE ETHANOL OF SPIRIT DRINKS OF VITIVINICULTURAL ORIGIN. (Type II)

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### 4. Principle

During photosynthesis, the assimilation of carbon dioxide by plants occurs according to 2 principle types of metabolism that are: metabolism  $\text{C}_3$  (Calvin cycle) and  $\text{C}_4$  (Hatch and Slack). These two means of photosynthesis present a different type of isotope fractionation. Products, such as sugars and alcohol, derived from  $\text{C}_4$  plants and fermentation, have higher levels of Carbon 13 than from  $\text{C}_3$  plants. Most plants, such as vines and sugar beet belong to the  $\text{C}_3$  group. Sugar cane and corn belong to the  $\text{C}_4$  group. Measuring the carbon 13 content enables the detection and the quantification of  $\text{C}_4$  (sugar cane or corn isoglucose) origin sugars which are added to products derived from grapes (grape musts, wines). The combined information on carbon 13 content and information obtained from SNIF-NMR enable the quantification of mixed sugars added or alcohol of plant origin  $\text{C}_3$  and  $\text{C}_4$ .

The carbon 13 content is determined on carbon dioxide resulting from the complete combustion of the sample. The abundance of the main mass isotopomers 44 ( $^{12}\text{C}^{16}\text{O}_2$ ), 45 ( $^{13}\text{C}^{16}\text{O}_2$  et  $^{12}\text{C}^{17}\text{O}^{16}\text{O}$ ) and 46 ( $^{12}\text{C}^{16}\text{O}^{18}\text{O}$ ), resulting from different possible combinations of isotopes  $^{18}\text{O}$ ,  $^{17}\text{O}$ ,  $^{16}\text{O}$ ,  $^{13}\text{C}$  et  $^{12}\text{C}$ , are determined from ion currents measured by three different collectors of isotopic mass spectrometers. The contributions of isotopomers  $^{13}\text{C}^{17}\text{O}^{16}\text{O}$  et  $^{12}\text{C}^{17}\text{O}_2$  can be neglected because of their low abundance. The ion current for  $m/z = 45$  is corrected for the contribution of  $^{12}\text{C}^{17}\text{O}^{16}\text{O}$  which is calculated according to the current intensity measured for  $m/z = 46$  while considering the relative abundance of  $^{18}\text{O}$  and  $^{17}\text{O}$  (Craig adjustment). The comparison with the calibrated reference and the international reference V-PDB enable the calculation of carbon 13 content on a relative scale of  $\delta^{13}\text{C}$ .

### 5. Reagents

The material and the consumables depend on the apparatus (6) used by the laboratory. The systems generally used are based on elementary analysers. These systems can be equipped to introduce the samples placed in sealed metal capsules or for the injection of liquid samples through a septum using a syringe.

Depending on the type of instrument used, the reference material, reagents, and consumables can be used:

# COMPENDIUM OF INTERNATIONAL METHODS OF ANALYSIS FOR SPIRITUOUS BEVERAGES AND ALCOHOLS

## DETERMINATION BY ISOTOPE RATIO MASS SPECTROMETRY OF THE $^{13}\text{C}/^{12}\text{C}$ RATIO OF WINE

### ETHANOL OF SPIRIT DRINKS OF VITIVINICULTURAL ORIGIN. (Type II)

- Reference material available from the IAEA:

| Name       | Material     | $\delta^{13}\text{C}$ versus V-PDB (‰) |
|------------|--------------|--|
| -IAEA-CH-6 | saccharose   | -10,4 ‰                                |
| -IAEA-CH-7 | polyethylene | -31,8 ‰                                |
| - NBS22    | oil          | -29,7 ‰                                |
| - USGS24   | graphite     | -16,1 ‰                                |

- Available from the IRMM in Geel (B) (Institut des Matériaux et Mesures de Référence) :

| Name      | Material                             | $\delta^{13}\text{C}$ versus V-PDB (‰) |
|-----------|--------------------------------------|--|
| - CRM 656 | Wine alcohol                         | -26,93 ‰                               |
| - CRM 657 | glucose                              | -10,75 ‰                               |
| - CRM 660 | Aqueous alcoholic solution (ABV 12%) | -26,72 ‰                               |

Standard working standard with a known  $^{13}\text{C}/^{12}\text{C}$  ratio calibrated with international reference materials.

A standard list of consumables established for continuous flow systems follows below:

- Helium for analysis (CAS 07440-59-7)
- Oxygen for analysis (CAS 07782-44-7)
- Carbon dioxide for analysis, used as a secondary reference gas for the content of carbon13 (CAS 00124-38-9)
- Oxidation reagent for the oven and the combustion system as follows: copper oxide (II) for elementary analysis (CAS 1317-38-0)
- Drying agent to eliminate water produced by combustion. For example: anhydron for elementary analysis (magnesium perchlorate) (CAS 10034-81-8).

*This is not necessary for apparatus equipped with a water elimination system by cryo-*

# COMPENDIUM OF INTERNATIONAL METHODS OF ANALYSIS FOR SPIRITUOUS BEVERAGES AND ALCOHOLS

## DETERMINATION BY ISOTOPE RATIO MASS SPECTROMETRY OF THE $^{13}\text{C}/^{12}\text{C}$ RATIO OF WINE

ETHANOL OF SPIRIT DRINKS OF VITIVINICULTURAL ORIGIN. (Type II)  
*trapping or through selective permeable capillaries.*

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### 6. Apparatus and materials

#### 6.1. Isotope ratio mass spectrometry (IRMS)

Isotope ratio mass spectrometry (IRMS) enables the determination of the relative contents of  $^{13}\text{C}$  of naturally occurring  $\text{CO}_2$  gas with an internal accuracy of 0.05‰ or expressed in relative value (9). Internal accuracy here is defined as the difference between 2 measurements of the same sample of  $\text{CO}_2$ . The mass spectrometer used to measure isotope ratios is generally equipped with a triple collector to simultaneously measure  $m/z = 44, 45$  and  $46$  intensities. The isotope ratio mass spectrometry must either be equipped with a double introduction system to alternately measure the unknown sample and a reference sample, or use an integrated system that carries out quantitative combustion on samples and separates the carbon dioxide from the other combustion products before measuring the mass spectrum.

#### 6.2. Combustion apparatus

Combustion apparatus able to quantitatively convert ethanol into carbon dioxide and capable of eliminating all other combustion products including water, without any isotopic fractionation. The apparatus can be either an integrated continuous flow system integrated with mass spectrometry (6.2.1), of an autonomous combustion system (6.2.2). The apparatus must be as precise as indicated in (11).

##### 6.2.1. Continuous flow systems

These consist of either an elemental analyser, or a gas chromatograph equipped with an online combustion system.

The following laboratory material is used for systems equipped for the introduction of samples contained in metallic capsules :

- volumetric micropipette with appropriate cones
- balance with 1  $\mu\text{g}$  accuracy or better
- tool for capsule sealing
- tin capsules for liquid samples
- tin capsules for solid samples

The following laboratory material is needed when using an elemental analyser

# COMPENDIUM OF INTERNATIONAL METHODS OF ANALYSIS FOR SPIRITUOUS BEVERAGES AND ALCOHOLS

## DETERMINATION BY ISOTOPE RATIO MASS SPECTROMETRY OF THE $^{13}\text{C}/^{12}\text{C}$ RATIO OF WINE

### ETHANOL OF SPIRIT DRINKS OF VITIVINICULTURAL ORIGIN. (Type II)

equipped with a liquid injector or in the case of a preparation system for combustion chromatography:

- syringe for liquids
- flasks equipped with sealed closing system and inert septa

The laboratory materials indicated in the lists are examples that are susceptible of being replaced by other equivalent performance material depending on the type of combustion apparatus and of mass spectrometry used by the laboratory.

#### 6.2.2. Separate preparation systems

In this case the samples of carbon dioxide resulting from the combustion of samples to be analyzed and the reference sample are collected in ampoules which are then put in a double entry spectrometer system to carry out isotopic analyses. Several types of combustion apparatus described in the literature can be used:

- Closed combustion system filled with circulating oxygen gas
- Elemental analyser with helium and oxygen flows
- Sealed glass ampoule filled with copper oxide (II) used as an oxidation agent

## 7. Preparation of samples for tests

Ethanol must be extracted from the spirit drink before isotopic testing. This is carried out by distilling the beverage as described in §3.1 of the SNIF-NMR method for the determination by NMR of the deuterium distribution in the ethanol of spirit drinks of vitivinicultural origin.

## 8. Procedure

All preparation steps must be carried out without any significant ethanol loss through evaporation, which would change the isotopic composition of the sample.

The description that follows makes reference to the procedure generally used for ethanol sample combustion using commercial automatic combustion systems. All other methods, ensuring that ethanol samples are converted quantitatively into carbon dioxide without the evaporation of ethanol, can use the preparation of carbon dioxide for isotopic analyses. An experimental procedure based on the use of an elemental analyser:

#### a) Placing the samples in capsules

- use capsules, tweezers and a clean preparation tray
- take an appropriate sized capsule using tweezers

# COMPENDIUM OF INTERNATIONAL METHODS OF ANALYSIS FOR SPIRITUOUS BEVERAGES AND ALCOHOLS

## DETERMINATION BY ISOTOPE RATIO MASS SPECTROMETRY OF THE $^{13}\text{C}/^{12}\text{C}$ RATIO OF WINE

### ETHANOL OF SPIRIT DRINKS OF VITIVINICULTURAL ORIGIN. (Type II)

- introduce an appropriate amount of liquid into the capsule using a micropipette
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Note: 3.84 mg of absolute ethanol or 4.17 mg of distillate with an alcohol content of 92 % m/m are necessary to obtain 2 mg of carbon. The appropriate quantity of distillate must be calculated in the same way according to the quantity of carbon necessary based on the mass spectrometer's sensitivity.

- close the capsule with the sealing tool.
  - each capsule must be completely sealed. If not, it must be discarded and a new capsule must be prepared.
  - two capsules must be prepared for every sample
  - place the capsules in an appropriate place on the elemental analyser sample tray. Every capsule must be carefully identified in order by number.
  - systematically place capsules containing working references at the beginning and the end of the sample series
  - regularly insert a check sample in the sample series.
- b) check and adjust the elemental analysis and mass spectrometry instrumentation
- adjust the temperature of the elemental analyzer ovens and the helium and oxygen gas flow for an optimal combustion of the sample;
  - check the elemental analysis system and the mass spectrometry system for leaks (for example by checking the ion current where  $m/z = 28$  corresponds to  $\text{N}_2$ );
  - adjust the mass spectrometer to measure the ion current intensities for  $m/z = 44, 45$  and  $46$ ;
  - check the system using known reference samples before starting to measure the samples.

c) To carry out a series of measurements

The samples that are placed on the autosampler of the elemental analyser or chromatograph are introduced successively. The carbon dioxide for each sample combustion is eluted into the mass spectrometer which measures the ion current. The data system records the ion current intensities and calculates the  $\delta$  values for each sample (9).

## 9. Calculation

# COMPENDIUM OF INTERNATIONAL METHODS OF ANALYSIS FOR SPIRITUOUS BEVERAGES AND ALCOHOLS

## DETERMINATION BY ISOTOPE RATIO MASS SPECTROMETRY OF THE $^{13}\text{C}/^{12}\text{C}$ RATIO OF WINE ETHANOL OF SPIRIT DRINKS OF VITIVINICULTURAL ORIGIN. (Type II)

The objective of the method is to measure the  $^{13}\text{C}/^{12}\text{C}$  isotopic ratio of the ethanol extracted from spirit drinks. The  $^{13}\text{C}/^{12}\text{C}$  isotopic ratio can be expressed by its deviation compared to the reference work. The carbon 13 ( $\delta^{13}\text{C}$ ) isotopic ratio is calculated on a delta scale per thousand by comparing the results obtained for the sample to be measured to the working reference calibrated previously based on the primary international reference (V-PDB). The  $\delta^{13}\text{C}$  values are expressed compared to the working reference:

$$\delta^{13}\text{C}_{\text{ech/ref}}\text{‰} = 1000 \times (R_{\text{ech}} - R_{\text{ref}})/R_{\text{ref}}$$

where  $R_{\text{ech}}$  and  $R_{\text{ref}}$  are respectively the isotopic ratio  $^{13}\text{C}/^{12}\text{C}$  of the sample and the working reference.

The  $\delta^{13}\text{C}$  values are thus expressed using V-PDB:

$$\delta^{13}\text{C}_{\text{ech/V-PDB}}\text{‰} = \delta^{13}\text{C}_{\text{ech/ref}} + \delta^{13}\text{C}_{\text{ref/V-PDB}} + \left( \delta^{13}\text{C}_{\text{ech/ref}} \times \delta^{13}\text{C}_{\text{ref/V-PDB}} \right) / 1000$$

where  $\delta^{13}\text{C}_{\text{ref/V-PDB}}$  is the isotopic deviation determined beforehand for the working reference to V-PDB.

Small variations may occur while measuring on line due to changes in the instrumental conditions. In this case the  $\delta^{13}\text{C}$  samples must be corrected according to the difference in the  $\delta^{13}\text{C}$  value from the working reference and the real value, which was calibrated beforehand against V-PDB by comparison with one of the international reference materials. Between two measurements of the working reference, the variation and therefore the correction applied to the sample results may be assumed to be linear. The working reference must be measured at the beginning and at the end of all sample series. A correction can then be calculated for each sample using linear interpolation between the two values (the difference between the assigned value of the working reference and the measurement values obtained).

### 10. Quality assurance and control

Check that the  $^{13}\text{C}$  value for the working reference does not differ by more than 0.5‰ of the accepted value. If not, the spectrometer must be checked and possibly readjusted.

For each sample, verify that the difference in the results between the 2 capsules measured successively is under 0.3 ‰. The final result for a given sample is the

# COMPENDIUM OF INTERNATIONAL METHODS OF ANALYSIS FOR SPIRITUOUS BEVERAGES AND ALCOHOLS

## DETERMINATION BY ISOTOPE RATIO MASS SPECTROMETRY OF THE $^{13}\text{C}/^{12}\text{C}$ RATIO OF WINE

### ETHANOL OF SPIRIT DRINKS OF VITIVINICULTURAL ORIGIN. (Type II)

average value of the 2 capsules. If the deviation is higher than 0.3 ‰ the measurement should be repeated.

Measurement condition monitoring can be based on the ion current intensity for  $m/z = 44$  and is proportional to the quantity of carbon injected in the elemental analyzer. Under standard conditions, the ion current intensity should be almost constant for the samples analysed. A significant deviation could be indicative of ethanol evaporation (an imperfect seal on a capsule), an instability of the elemental analyser, or the mass spectrometer.

#### 11. Method performance characteristics (precision)

One collaborative analysis (11.1) was carried out on distillates containing alcohol of vinous origin, and cane and beet alcohol, in addition to different mixtures of these three origins. This study did not take into account the distillation step, and further information from other joint laboratory studies on wine (11.2) and in particular proficiency testing (11.3) for isotopic measurements have also been considered. The results show that different distillation systems under satisfactory conditions, and in particular those used for SNIF-NMR measurements, do not have significant variability for determining the  $\delta^{13}\text{C}$  of ethanol in wine. It is reasonable to suppose that this would likewise be true for the ethanol of spirit drinks. The precision parameters observed for wine are almost identical to those obtained in the joint study on distillates (11.1).

##### 11.1. Joint study on distillates

|                                |                                      |              |                    |
|--------------------------------|--------------------------------------|--------------|--------------------|
| Year of inter laboratory study | 1996                                 |              |                    |
| Number of laboratories         | 20                                   |              |                    |
| Number of samples              | 6 samples in double-blind comparison |              |                    |
| Analysis                       | ethanol $\delta^{13}\text{C}$        |              |                    |
| Sample code                    | Vinous origin alcohol                | Beet alcohol | Sugar cane alcohol |
| A & G                          | 80%                                  | 10%          | 10%                |
| B & C                          | 90%                                  | 10%          | 0%                 |
| D & F                          | 0%                                   | 100%         | 0%                 |



# COMPENDIUM OF INTERNATIONAL METHODS OF ANALYSIS FOR SPIRITUOUS BEVERAGES AND ALCOHOLS

## DETERMINATION BY ISOTOPE RATIO MASS SPECTROMETRY OF THE $^{13}\text{C}/^{12}\text{C}$ RATIO OF WINE

### ETHANOL OF SPIRIT DRINKS OF VITIVINICULTURAL ORIGIN. (Type II)

| ETHANOL OF SPIRIT DRINKS OF VITIVINICULTURAL ORIGIN. (Type II)      |        |        |        |        |        |        |
|---|--------|--------|--------|--------|--------|--------|
| E & I   | 90%    |        | 0%     |        | 10%    |        |
| H & K   | 100%   |        | 0%     |        | 0%     |        |
| J & L   | 0%     |        | 0%     |        | 100%   |        |
| Samples   | A / G  | B / C  | D / F  | E / I  | H / K  | J / L  |
| Number of laboratories retained after eliminating anomalous results | 19     | 18     | 17     | 19     | 19     | 19     |
| Number of results accepted  | 38     | 36     | 34     | 38     | 38     | 38     |
| Average value ( $\delta^{13}\text{C}$ ) ‰                           | -25.32 | -26.75 | -27.79 | -25.26 | -26.63 | -12.54 |
| $\text{Sr}^2$   | 0.0064 | 0.0077 | 0.0031 | 0.0127 | 0.0069 | 0.0041 |
| Repeatability standard deviation (Sr) ‰                             | 0.08   | 0.09   | 0.06   | 0.11   | 0.08   | 0.06   |
| Repeatability limit r (2.8×S <sub>r</sub> ) ‰                       | 0.22   | 0.25   | 0.16   | 0.32   | 0.23   | 0.18   |
| $\text{S}_\text{R}^2$   | 0.0389 | 0.0309 | 0.0382 | 0.0459 | 0.0316 | 0.0584 |
| Reproducibility standard deviation (S <sub>R</sub> ) ‰              | 0.20   | 0.18   | 0.20   | 0.21   | 0.18   | 0.24   |
| Reproducibility limit R (2,8× S <sub>R</sub> ) ‰                    | 0.55   | 0.9    | 0.55   | 0.60   | 0.50   | 0.68   |

#### 11.2. Inter laboratory study on two wines and one alcohol

|                                 |      |
|---------------------------------|------|
| Year of inter laboratory trial: | 1996 |
|---------------------------------|------|

# COMPENDIUM OF INTERNATIONAL METHODS OF ANALYSIS FOR SPIRITUOUS BEVERAGES AND ALCOHOLS

## DETERMINATION BY ISOTOPE RATIO MASS SPECTOMETRY OF THE $^{13}\text{C}/^{12}\text{C}$ RATIO OF WINE

### ETHANOL OF SPIRIT DRINKS OF VITIVINICULTURAL ORIGIN. (Type II)

|  |  |            |         |
|--|--|------------|---------|
| Number of laboratories                         | 14 for distillation of wine and 7 for also measuring the $\delta^{13}\text{C}$ of ethanol in wine<br>8 for measuring the $\delta^{13}\text{C}$ in the alcohol sample |            |         |
| Number of samples                              | 3 (White wine ABV 9.3 % vol., white wine ABV 9.6 % and alcohol of strength 93% m/m)  |            |         |
| Analysis                                       | ethanol $\delta^{13}\text{C}$  |            |         |
| Samples  | Red wine   | White wine | Alcohol |
| Number of laboratories                         | 7  | 7          | 8       |
| Number of accepted results                     | 7  | 7          | 8       |
| Average value ( $\delta^{13}\text{C}$ ) ‰      | -26.20   | -26.20     | -25.08  |
| Reproducibility variance $S_r^2$               | 0.0525   | 0.0740     | 0.0962  |
| Reproducibility standard deviation ( $S_r$ ) ‰ | 0.23   | 0.27       | 0.31    |
| Reproducibility limit R ( $2.8 \times S_r$ ) ‰ | 0.64   | 0.76       | 0.87    |

Different distillation systems were used by the participating laboratories. The  $\delta^{13}\text{C}$  isotopic determinations carried out in one laboratory on all of the distillates returned by the participants, did not reveal any anomalous values or significantly distinct average values. The variation in results ( $S^2 = 0.0059$ ) is comparable to repeatability variances  $S_r^2$  from the collaborative study on distillates (11.1).

#### 11.3. Results from proficiency-testing studies

Since December 1994 international proficiency testing to determine the isotopic measurements for wine and alcohol (ABV of distillates 96 % vol.) have been organized regularly. The results enable participating laboratories to check the quality of their analyses. Statistical results enable an appreciation of the variability of determinations under reproducibility conditions. This enables an estimation of the variance parameters and the reproducibility limit. The results obtained for the wine and ethanol distillate  $\delta^{13}\text{C}$  determinations are summarized in the table below:

# COMPENDIUM OF INTERNATIONAL METHODS OF ANALYSIS FOR SPIRITUOUS BEVERAGES AND ALCOHOLS

## DETERMINATION BY ISOTOPE RATIO MASS SPECTROMETRY OF THE $^{13}\text{C}/^{12}\text{C}$ RATIO OF WINE ETHANOL OF SPIRIT DRINKS OF VITIVINICULTURAL ORIGIN. (Type II)

| Wines            |    |       |       |      | Distillates |       |       |      |
|------------------|----|-------|-------|------|-------------|-------|-------|------|
| Date             | N  | SR    | S2R   | R    | N           | SR    | S2R   | R    |
| Dec. 1994        | 6  | 0,210 | 0,044 | 0,59 | 6           | 0,151 | 0,023 | 0,42 |
| June 1995        | 8  | 0,133 | 0,018 | 0,37 | 8           | 0,147 | 0,021 | 0,41 |
| Dec. 1995        | 7  | 0,075 | 0,006 | 0,21 | 8           | 0,115 | 0,013 | 0,32 |
| March 1996       | 9  | 0,249 | 0,062 | 0,70 | 11          | 0,278 | 0,077 | 0,78 |
| June 1996        | 8  | 0,127 | 0,016 | 0,36 | 8           | 0,189 | 0,036 | 0,53 |
| Sept. 1996       | 10 | 0,147 | 0,022 | 0,41 | 11          | 0,224 | 0,050 | 0,63 |
| Dec. 1996        | 10 | 0,330 | 0,109 | 0,92 | 9           | 0,057 | 0,003 | 0,16 |
| March 1997       | 10 | 0,069 | 0,005 | 0,19 | 8           | 0,059 | 0,003 | 0,16 |
| June 1997        | 11 | 0,280 | 0,079 | 0,78 | 11          | 0,175 | 0,031 | 0,49 |
| Sept 1997        | 12 | 0,237 | 0,056 | 0,66 | 11          | 0,203 | 0,041 | 0,57 |
| Dec. 1997        | 11 | 0,127 | 0,016 | 0,36 | 12          | 0,156 | 0,024 | 0,44 |
| March 1998       | 12 | 0,285 | 0,081 | 0,80 | 13          | 0,245 | 0,060 | 0,69 |
| June 1998        | 12 | 0,182 | 0,033 | 0,51 | 12          | 0,263 | 0,069 | 0,74 |
| Sept 1998        | 11 | 0,264 | 0,070 | 0,74 | 12          | 0,327 | 0,107 | 0,91 |
| Weighted average |    | 0,215 | 0,046 | 0,60 |             | 0,209 | 0,044 | 0,59 |

N: number of participating laboratorie

### 12. Bibliography

# COMPENDIUM OF INTERNATIONAL METHODS OF ANALYSIS FOR SPIRITUOUS BEVERAGES AND ALCOHOLS

## DETERMINATION BY ISOTOPE RATIO MASS SPECTROMETRY OF THE $^{13}\text{C}/^{12}\text{C}$ RATIO OF WINE

### ETHANOL OF SPIRIT DRINKS OF VITIVINICULTURAL ORIGIN. (Type II)

- 
- Determination by NMR of the distribution of deuterium in ethanol of spirit drinks of vitivinicultural origin
  - OIV Compendium of international methods of analysis of spirit drinks of vitivinicultural origin.
  - Determination by isotopic mass spectrometry of the  $^{13}\text{C}/^{12}\text{C}$  isotopic ratio of wine ethanol or ethanol obtained from fermentation of musts, concentrated musts or grape sugar
  - OIV Compendium of International Methods of Analysis of Wine and Musts. Detecting enrichment of musts, concentrated musts, grape and wine sugars by application of nuclear magnetic resonance of deuterium (RMN-FINS/SNIF-NMR)
  - E.C. Regulation. Community analytical methods which can be applied in the wine sector, N°.2676/90. Detecting enrichment of grape musts, concentrated grape musts, rectified concentrated grape musts and wines by application of nuclear magnetic resonance of deuterium (SNIF-NMR)
  - *Official Journal of the European Communities*, NoL 272, Vol 33, 64-73, 3 October 1990
  - Inter-laboratory study about the determination of  $\delta^{13}\text{C}$  in wine ethanol OIV FV No 1051
  - Fidélité de la détermination du rapport isotopique  $^{13}\text{C}/^{12}\text{C}$  de l'éthanol du vin OIV FV No 1116.
  - Stable carbon isotope content in ethanol of EC data bank wines from Italy, France and Germany. A Rossmann ; H-L Schmidt ; F. Reniero ; G. Versini ; I. Moussa ; M.- H. Merle. Z. Lebensm. Unters. Forsch., 1996, 203, PP. 293-301.